

***N*-(1-Cyano-1-methylbutyl)-1,2,3-benzothiadiazole-7-carboxamide**Ying-Wei Ai,^a Feng-Li Liu,^b
Zhi-Jin Fan,^{b*} Hai-Bin Song^b and
Kai-Sheng Nie^b^aCollege of Life Science, Sichuan University, Chengdu 610065, People's Republic of China, and ^bState Key Laboratory of Elemento-Organic Chemistry, Nankai University, Tianjin 300071, People's Republic of China

Correspondence e-mail: fanzj@nankai.edu.cn

Key indicatorsSingle-crystal X-ray study
T = 294 K
Mean $\sigma(\text{C}-\text{C}) = 0.005 \text{ \AA}$
R factor = 0.040
wR factor = 0.107
Data-to-parameter ratio = 14.2For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.

In the title compound, $\text{C}_{13}\text{H}_{14}\text{N}_4\text{OS}$, which was synthesized as a candidate plant activator by the reaction of 1-amino-1-methyl-1-propylacetonitrile and 1,2,3-benzothiadiazole-7-carbonyl chloride, the benzothiadiazole moiety is essentially planar, forming a dihedral angle of $14.5(4)^\circ$ with the amide group. In the crystal structure, molecules are linked by intermolecular $\text{N}-\text{H}\cdots\text{O}$ hydrogen bonds [$\text{H}\cdots\text{O} = 2.20(4) \text{ \AA}$] to form extended chains in the *a*-axis direction.

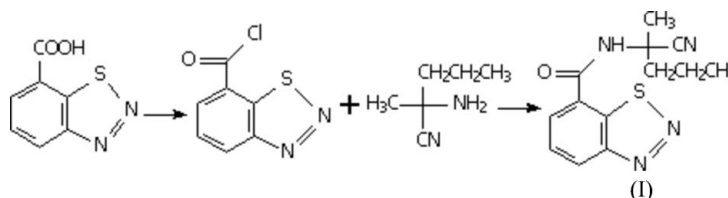
Received 28 September 2005

Accepted 24 October 2005

Online 27 October 2005

Comment

Derivatives of acibenzolar-*S*-methyl or BTH (*S*-methyl 1,2,3-benzothiadiazole-7-carbothioate) possess systemic acquired resistance against fungi and virus in agricultural practice. They are potential plant activators with environmentally friendly characteristics (Gozzo, 2003). Many studies have modified the structure (Bao, Liu *et al.*, 2005) in order to find more powerful plant activator candidates that can be applied to the prevention and cure of plant diseases and viruses, especially the tobacco mosaic virus (TMV). A search of the Cambridge Structural Database (Version 5.26 with updates to August 2005; Allen, 2002) revealed only seven structures containing the benzothiadiazole group and we have recently determined the structures of three more (Bao, Fan *et al.*, 2005; Liu *et al.*, 2005; Zhao *et al.*, 2005). We have now determined the structure of the title compound, (I), to investigate its quantitative structure–activity relationship (QSAR).



The molecular structure of (I) is shown in Fig. 1. The benzothiadiazole moiety is essentially planar (r.m.s. deviation 0.011 \AA), forming a dihedral angle with the C7, O1 and N3 plane of atoms of $14.5(4)^\circ$. The approximate coplanarity of the amine and benzothiadiazole groups, in conjunction with the observed bond lengths and angles (Table 1), suggests that an extended π -conjugated system exists in this part of the molecule.

In the crystal structure of (I), extended one-dimensional chains are formed *via* intermolecular $\text{N}-\text{H}\cdots\text{O}$ hydrogen bonds (Table 2).

Experimental

1,2,3-Benzothiadiazole-7-carboxylic acid was synthesized according to the description of Fan *et al.* (2005). The title compound was

prepared by reacting 1-amino-1-methyl-1-propylacetonitrile with 1,2,3-benzothiadiazole-7-carbonyl chloride (Bao, Fan *et al.*, 2005). The product obtained after silica-gel column chromatography was recrystallized from a mixture of petroleum ether (333–363 K) and ethyl acetate (2:1 *v/v*) at room temperature to yield colourless crystals suitable for X-ray diffraction analysis.

Crystal data

C₁₃H₁₄N₄OS
M_r = 274.34
 Monoclinic, *P*2₁
a = 6.236 (2) Å
b = 13.263 (4) Å
c = 8.667 (3) Å
 β = 110.740 (5)°
V = 670.4 (4) Å³
Z = 2

D_x = 1.359 Mg m⁻³
 Mo *K*α radiation
 Cell parameters from 2025 reflections
 θ = 2.5–24.6°
 μ = 0.24 mm⁻¹
T = 294 (2) K
 Block, colourless
 0.32 × 0.24 × 0.20 mm

Data collection

Bruker SMART-CCD area-detector diffractometer
 φ and ω scans
 Absorption correction: multi-scan (SADABS; Sheldrick, 1996)
T_{min} = 0.920, *T_{max}* = 0.953
 3879 measured reflections

2534 independent reflections
 2262 reflections with *I* > 2σ(*I*)
R_{int} = 0.022
 θ_{max} = 26.4°
h = -7 → 7
k = -13 → 16
l = -7 → 10

Refinement

Refinement on *F*²
R [*F*² > 2σ(*F*²)] = 0.040
wR (*F*²) = 0.107
S = 1.07
 2534 reflections
 179 parameters
 H atoms treated by a mixture of independent and constrained refinement

$w = 1/[\sigma^2(F_o^2) + (0.0645P)^2]$
 where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{max} < 0.001$
 $\Delta\rho_{max} = 0.18 \text{ e \AA}^{-3}$
 $\Delta\rho_{min} = -0.18 \text{ e \AA}^{-3}$
 Absolute structure: Flack (1983), with 1097 Friedel pairs
 Flack parameter: 0.03 (12)

Table 1 Selected geometric parameters (Å, °).

S1–C2	1.702 (3)	N3–C8	1.467 (4)
S1–N1	1.703 (4)	O1–C7	1.222 (4)
N1–N2	1.278 (5)	C1–C6	1.380 (4)
N2–C3	1.377 (5)	C1–C7	1.494 (4)
N3–C7	1.344 (4)	C2–C3	1.393 (5)
C2–S1–N1	92.29 (17)	O1–C7–C1	120.1 (3)
C7–N3–C8	123.0 (3)	N3–C7–C1	116.9 (3)
O1–C7–N3	122.9 (3)		
C13–C8–C10–C11	174.1 (3)	C8–C10–C11–C12	173.7 (3)

Table 2 Hydrogen-bond geometry (Å, °).

<i>D</i> –H... <i>A</i>	<i>D</i> –H	H... <i>A</i>	<i>D</i> ... <i>A</i>	<i>D</i> –H... <i>A</i>
N3–H3...N4 ⁱ	0.89 (4)	2.20 (4)	3.083 (4)	173 (3)

Symmetry code: (i) *x* + 1, *y*, *z*.

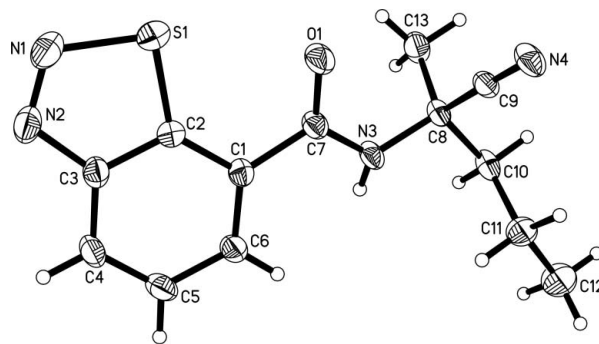


Figure 1 A view of (I), with displacement ellipsoids drawn at the 30% probability level. H atoms are shown as spheres of arbitrary radii.

The position of the amine H atom was refined independently with an isotropic displacement parameter. All H atoms bonded to C atoms were placed in idealized positions and constrained to ride on their parent atoms, with C–H distances ranging from 0.93 to 0.97 Å, and with *U_{iso}*(H) = 1.2*U_{eq}*(C), or 1.5*U_{eq}*(C_{methyl}).

Data collection: SMART (Bruker, 1998); cell refinement: SAINT (Bruker, 1999); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: SHELXTL (Bruker, 1999); software used to prepare material for publication: SHELXTL.

This study was funded in part by grants from the National Basic Research Programme of China (973 Programme) (grant No. 2003CB114402), the National Natural Science Foundation of China (grant No. 30270883) and the Key Laboratory of Pesticide Chemistry and Application Technology of the Ministry of Agriculture of China.

References

Allen F. H. (2002). *Acta Cryst.* **B58**, 380–388.
 Bao, L.-L., Fan, Z.-J., Song, H.-B & Kaisheng, N. (2005). *Acta Cryst.* **E61**, o3817–o3818.
 Bao, L.-L., Liu, F.-L. & Fan, Z.-J. (2005). *Chin. J. Pestic. Sci.* **7**, 201–209. (In Chinese).
 Bruker (1998). SMART. Bruker AXS Inc., Madison, Wisconsin, USA.
 Bruker (1999). SAINT and SHELXTL. Bruker AXS Inc., Madison, Wisconsin, USA.
 Fan, Z.-J., Liu, F.-L. & Liu, X.-F. (2005). Chin. Patent CN1 680 342A. (In Chinese).
 Flack, H. D. (1983). *Acta Cryst.* **A39**, 876–881.
 Gozzo, F. (2003). *J. Agric. Food. Chem.* **51**, 4487–4503.
 Liu, F.-L., Fan, Z.-J., Song, H.-B., Liu, X.-F. & Zhang, Y.-G. (2005). *Acta Cryst.* **E61**. Submitted.
 Sheldrick, G. M. (1996). SADABS. University of Göttingen, Germany.
 Sheldrick, G. M. (1997). SHELXS97 and SHELXL97. University of Göttingen, Germany.
 Zhao, J.-Z., Liu, F.-L., Song, H.-B & Fan, Z.-J. (2005). *Acta Cryst.* **E61**. Submitted.